

CLASS XII CHEMISTRY CH 2 SOLUTIONS



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2 Section-B 6. Calculate the freezing point of a solution containing 8.1 g of HBr in 100 g of water, assuming the acid to be 90 % ionized. [Given: Molar mass Br = 80 g/mol, K_f water = 1.86 K kg / mol]

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16. 18. 19. 20. 56/1/1 t I Èfiyà I Chlorobenzene is extremely less reactive towards a nucleophilic substitution reaction. Give two reasons for the same.

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Class XII: Chemistry Chapter 7: The p-Block Elements Top

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History of chemistry - Wikipedia

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Uranium hexafluoride - Wikipedia

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